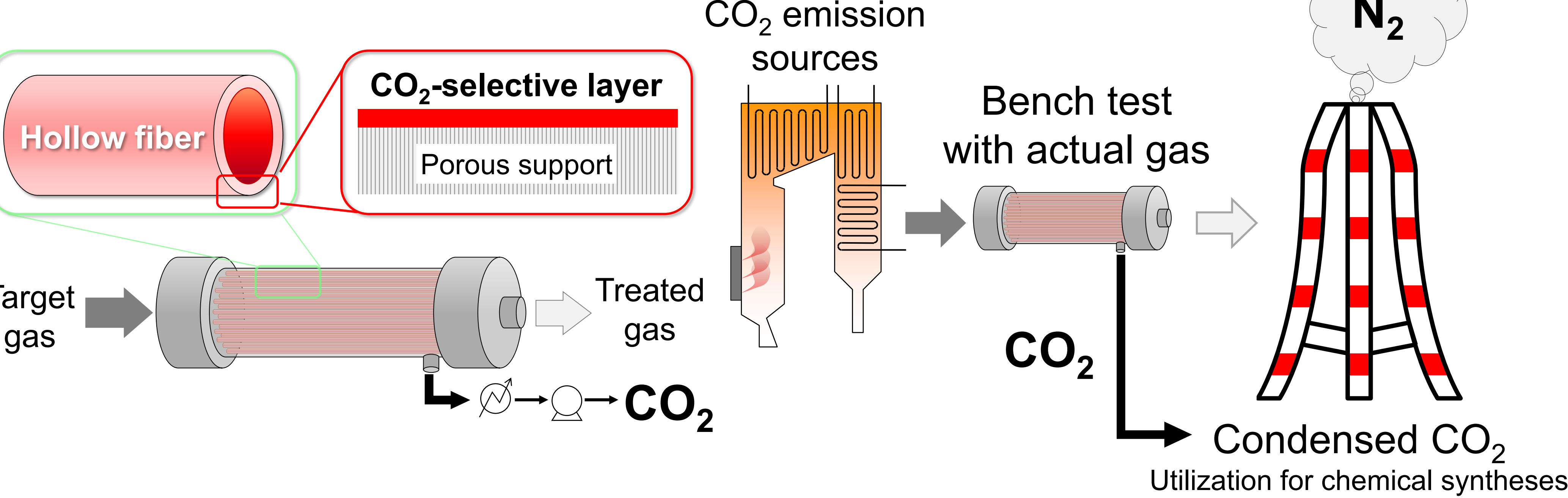


## Introduction

The increase in atmospheric CO<sub>2</sub> concentration leading to the Global Warming and Climate Change is one of the serious issues, and the urgent transition to a decarbonized society is a global challenge. To achieve this, CO<sub>2</sub> capture is now required from all sources of CO<sub>2</sub> emissions. The chemical absorption with aqueous amines has been employed for CO<sub>2</sub> capture in large-scale CO<sub>2</sub> emission sources, such as thermal power stations, and that requires significant energy. In addition, the technology is not suitable for the medium- and small-scale emission sources. Therefore, membrane separation is being recognized as an effective alternative CO<sub>2</sub> capture technology.

In this research group, amine-containing polymeric membranes have been investigated for CO<sub>2</sub> capture, where amines are physically immobilized in a polymer matrix. The obtained membranes demonstrated excellent CO<sub>2</sub> separation performance. The mechanism of preferential CO<sub>2</sub> permeation was also elucidated at the molecular level.

## Outline of the project

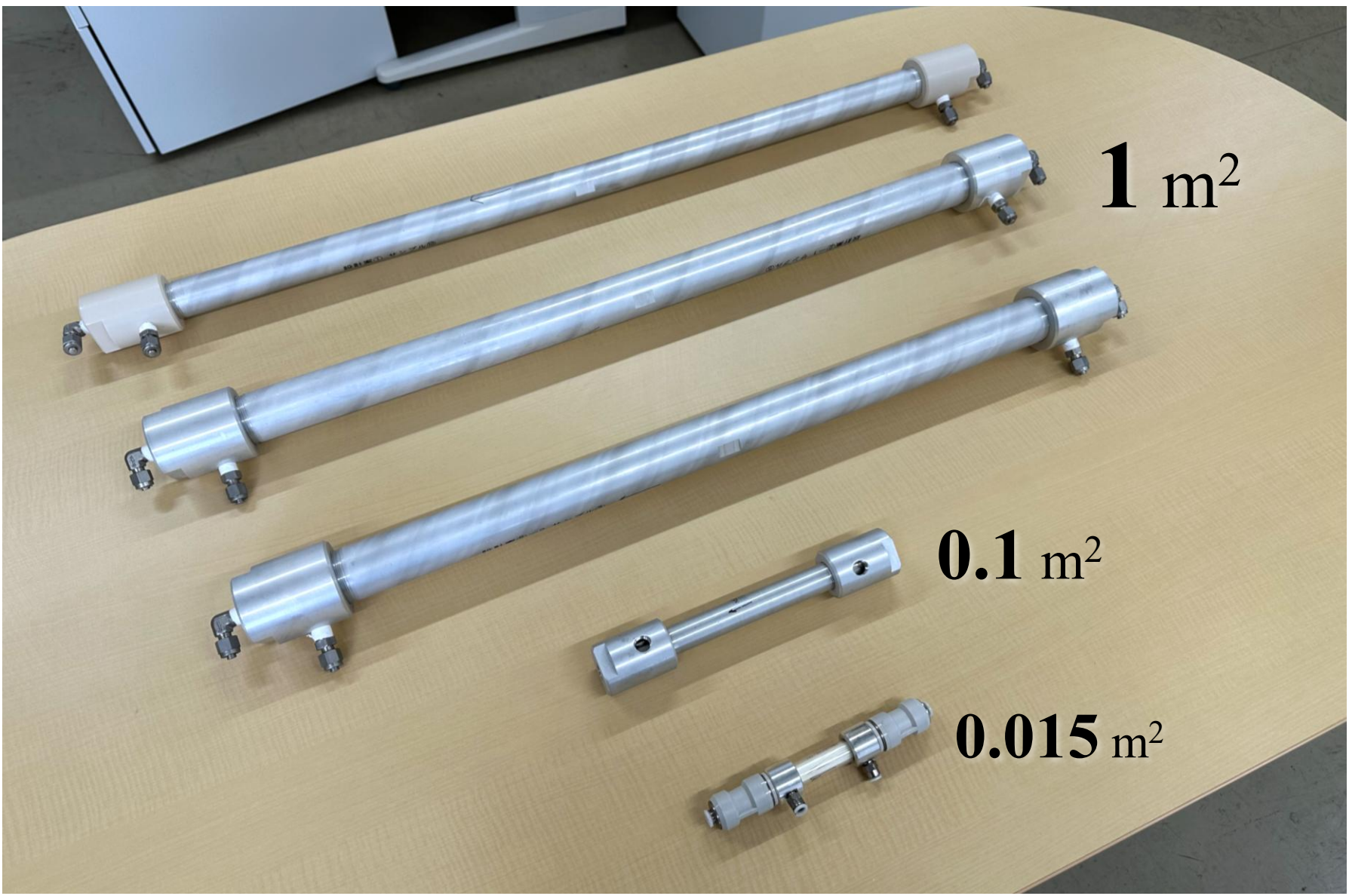


- [1] Preparation of CO<sub>2</sub> separation membrane modules
- [2] Bench-scale demonstration
- [3] Optimization of membrane separation process

Capture cost:  $\leq \$10/\text{t-CO}_2$

NEDO

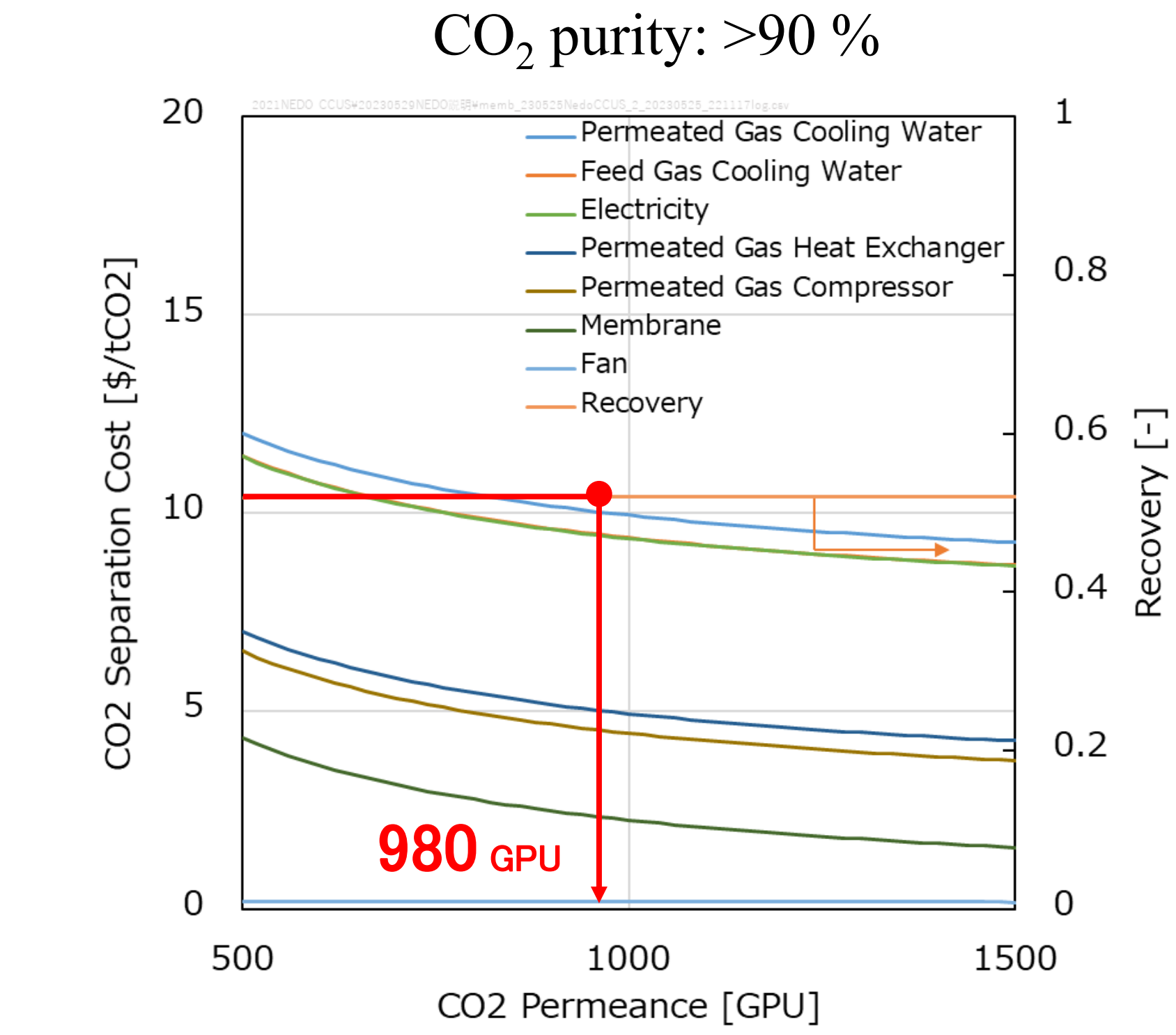
## Membrane modules prepared



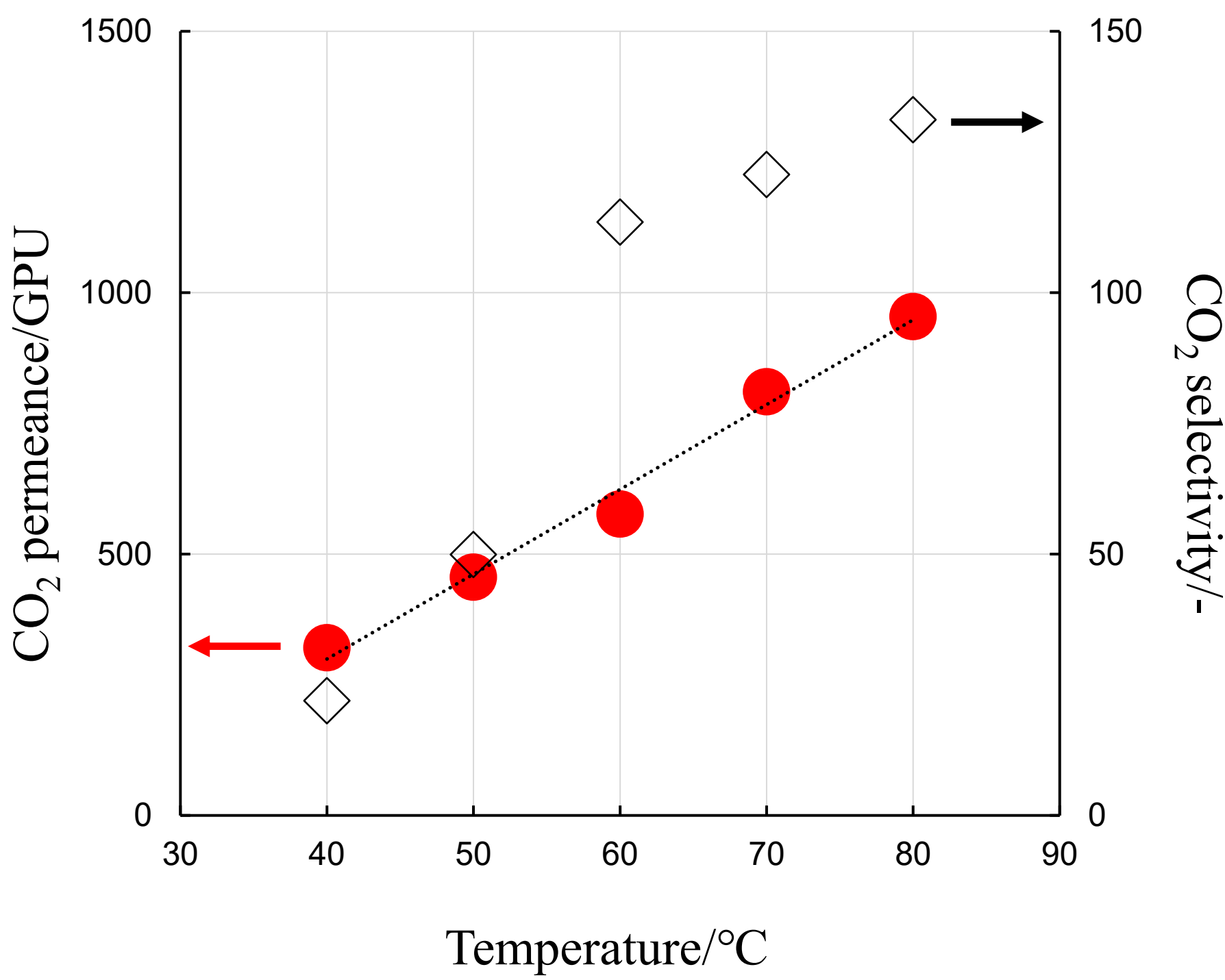
### Significant cost reductions in CO<sub>2</sub> capture

- Extremely simple preparation
- Easy mass production
- Convenient large-area production

## Cost analysis

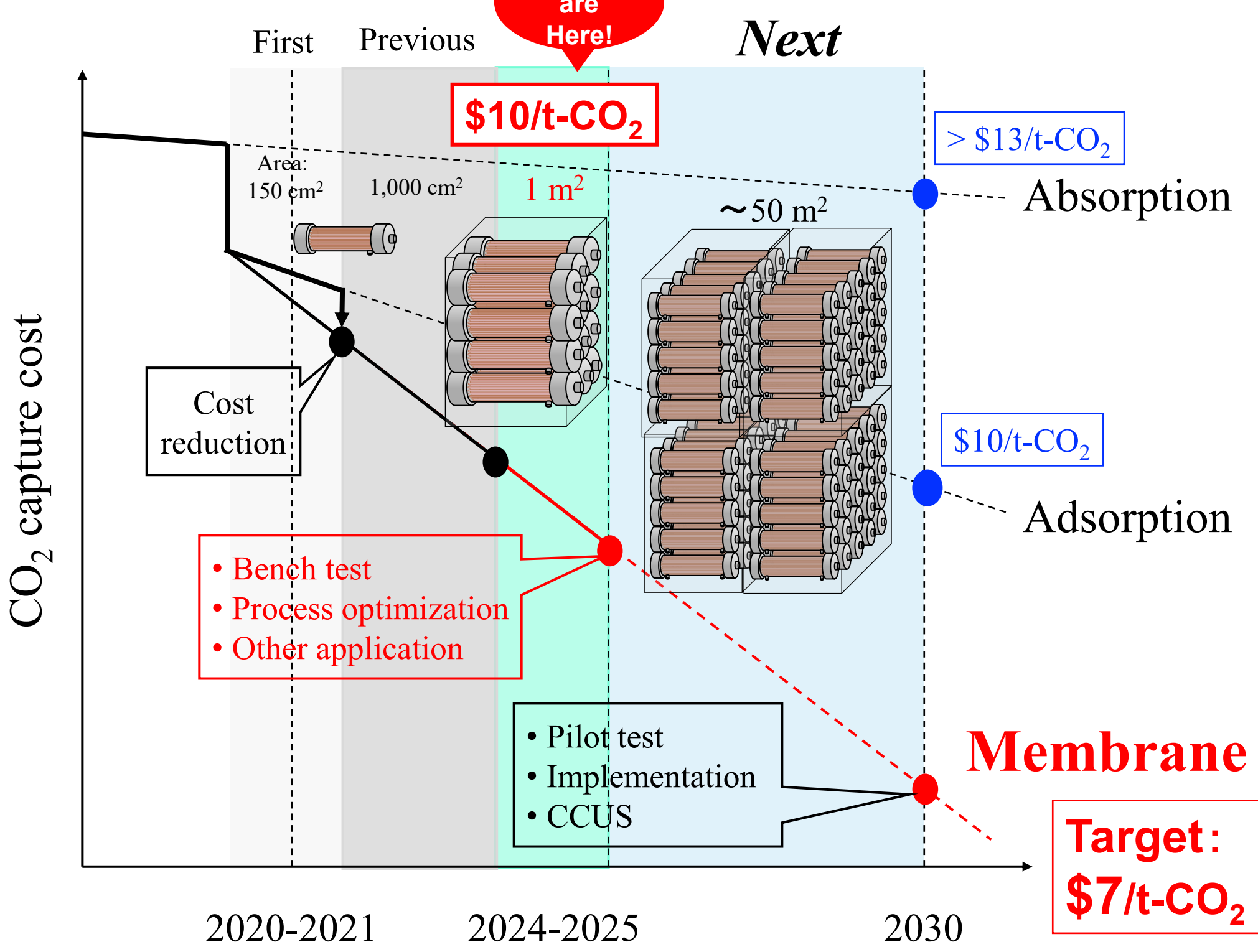


## Effect of temperature



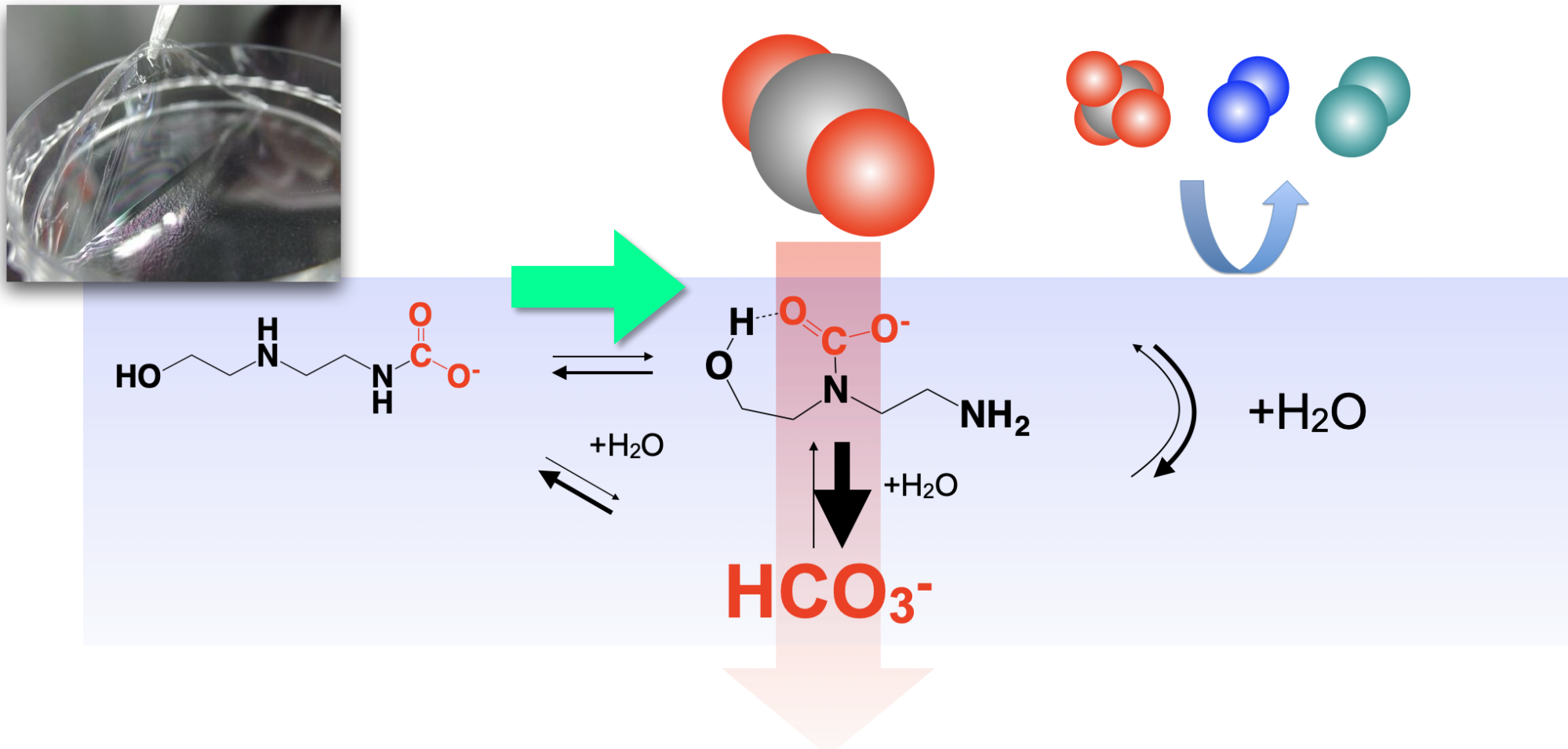
Membrane area: 150 cm<sup>2</sup>; Feed gas: CO<sub>2</sub>/N<sub>2</sub>=10/90 vol%; Relative humidity: 90 %

## R&D schedule



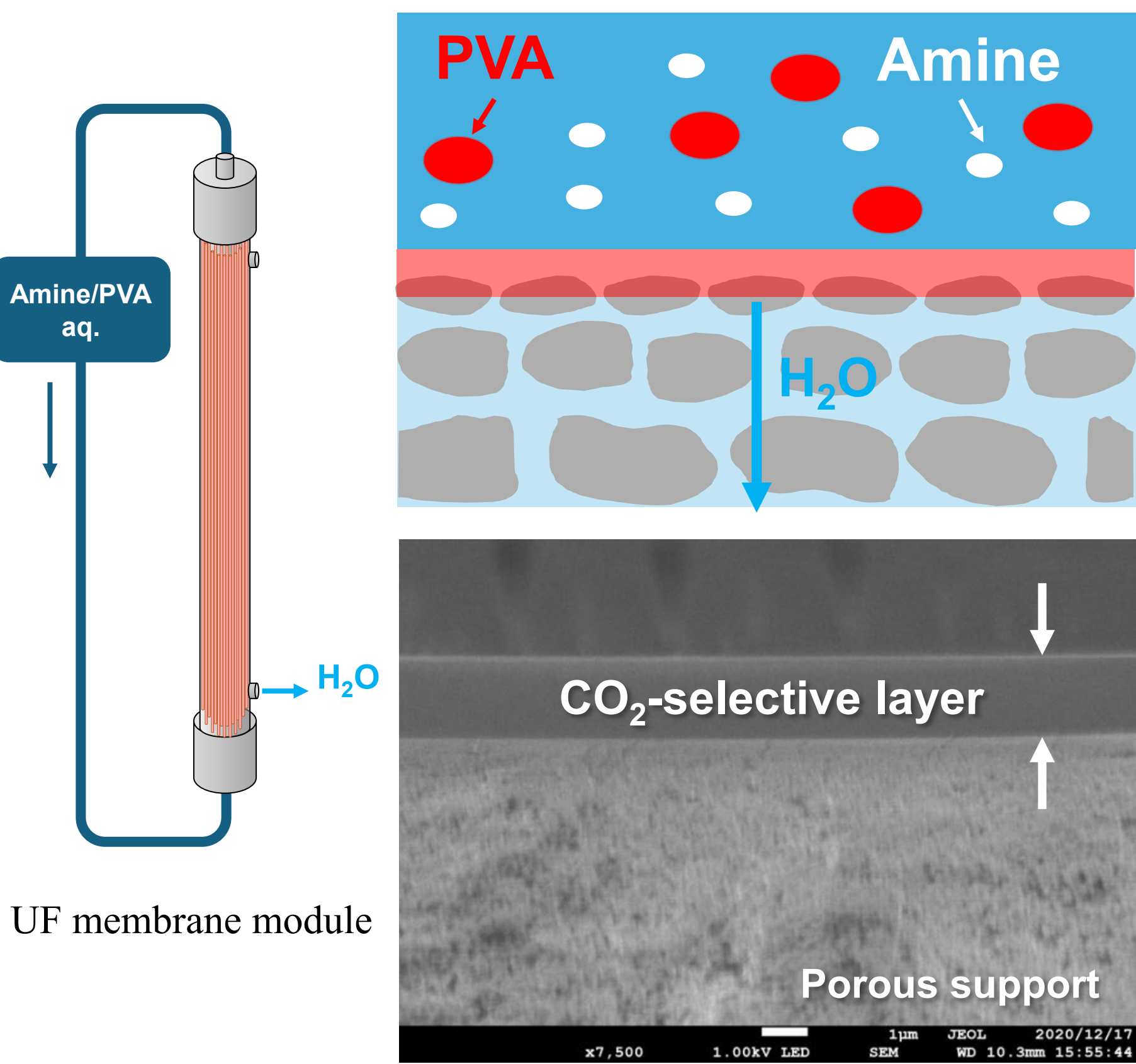
## Preferential CO<sub>2</sub> permeation

### Mechanism of gas transportation across amin-containing polymeric membranes

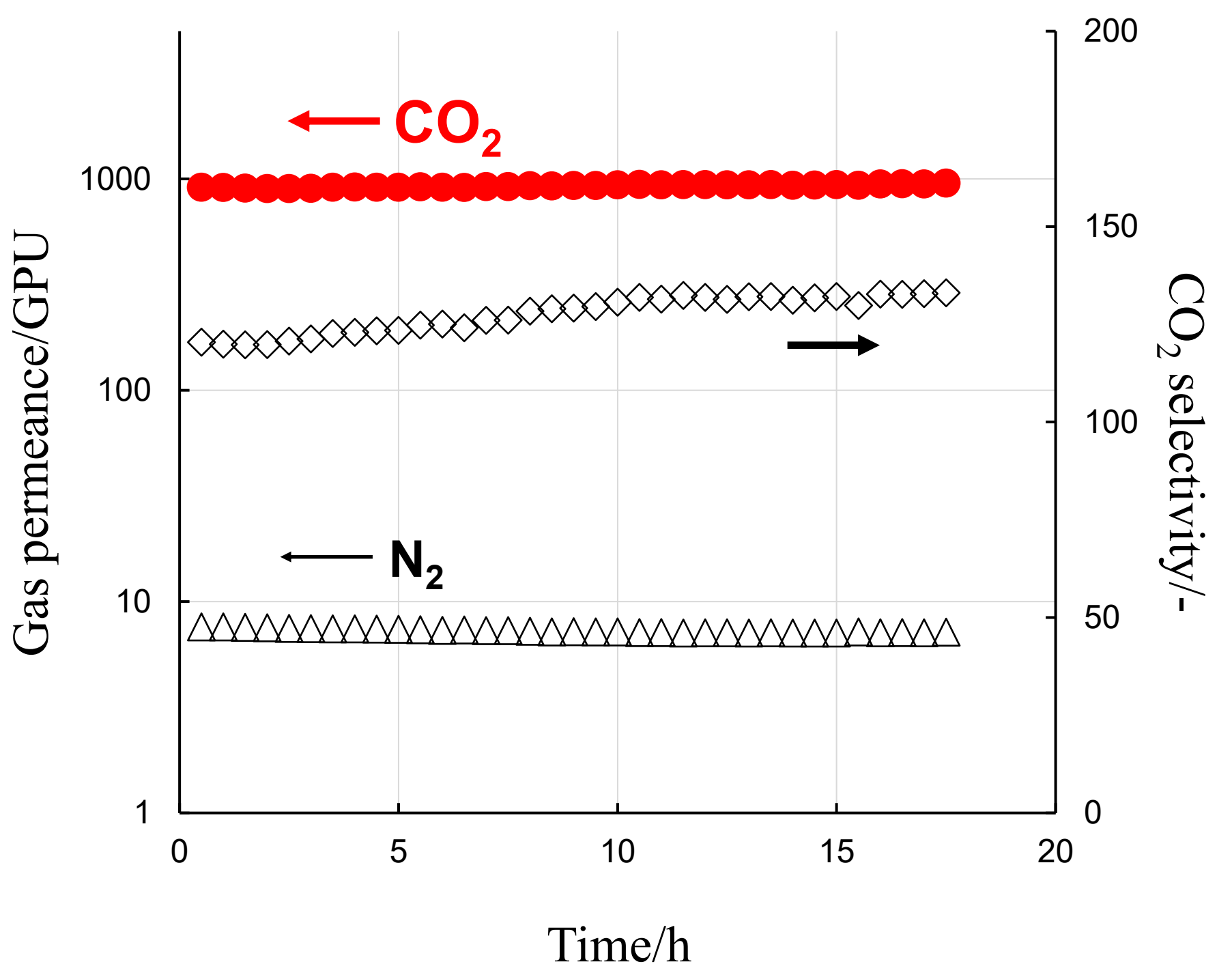


- CO<sub>2</sub> interacts to AEAE with the formation of carbamates.
- Hydroxyl group stabilizes the carbamate on secondary amine.
- Bicarbonate ion is released upon hydrolysis of the carbamate.
- Other gas permeation is suppressed by salting-out to give high CO<sub>2</sub> selectivity

## Membrane module preparation



## Time course



Membrane area: 150 cm<sup>2</sup>; Feed gas: CO<sub>2</sub>/N<sub>2</sub>=10/90 vol%; Relative humidity: 90 % at 80 °C

## Summary & future perspectives

The results of process simulation showed that the developed hollow fiber membrane module with a membrane area of 150 cm<sup>2</sup> has excellent CO<sub>2</sub> separation performance, and that the CO<sub>2</sub> separation and recovery cost is \$10/t-CO<sub>2</sub> through a bench test using actual exhaust gas. The membrane area was increased to 1 m<sup>2</sup> (20 cm<sup>2</sup>).

A bench test with a scaled-up membrane area of 1 m<sup>2</sup> (20 kg-CO<sub>2</sub>/day) is currently under consideration, and the membrane area will be further expanded to 50 m<sup>2</sup> in the next pilot test. The next pilot test will further increase the membrane area to 50 m<sup>2</sup>, with the aim of commercialization in 2030.

Given the membrane's proven efficacy, its implementation in other CO<sub>2</sub> emission sources, such as steel and cement facilities, is being contemplated.